

POLYSACCHARIDES OF BROWN SEAWEEDS

VIII. THE STRUCTURE OF THE SIDE CHAINS OF THE SARGASSAN MOLECULE

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UDC 547.917+ 639.64

The results of a study of sargassan, a glucuronoglycan from *Sargassum pallidum* (brown seaweed) [1], have shown that its molecule is based on a linear carbohydrate chain of alternating glucuronic acid and mannose residues [2]. We have attempted to determine individual features of the structure of the side chains of the sargassan molecule by analyzing the oligosaccharide fragments formed on partial hydrolysis with 1% sulfuric acid or 0.5 M oxalic acid by a previously described method [2]. The mixture of oligo- and monosaccharides was fractionated on Bio-Gel P-6 (elution with water), then by column chromatography on cellulose (with water-saturated butanol), and, finally, by preparative paper chromatography in the butanol-ethanol-water (40:11:19) and butan-1-ol-acetic acid-water (4:5:1, upper layer) systems. Six chromatographically individual oligosaccharides, A-F, were isolated (Table 1). The results of acid hydrolysis of the compounds obtained and of their tetrahydroborate-reduced analogs showed the monosaccharide compositions and reducing terminal residues of all the oligosaccharides (see Table 1). The quantitative monosaccharide compositions (by gas-liquid chromatography of the acetates of the aldonitriles [3]) and chromatographic mobility showed that the oligosaccharides A-D are disaccharides, while oligosaccharide E is apparently a trisaccharide.

Preliminary results obtained by means of color reactions with TTC [4] and with the diphenylamine-aniline reagent [5] indicate a 1,2-bond at the reducing end of oligosaccharides B, C, E, and F, a 1,3 bond in disaccharide A, and a 1,4 bond in disaccharide D. In addition, on alkaline hydrolysis [6], disaccharide A decomposed into its constituent monosaccharides, while the other oligosaccharides remained unchanged. As is well known, this shows that disaccharide A has a 1,3 bond.

The higher proportion of disaccharides A and D in the hydrolyzate from sargassan permitted them to be isolated in amounts sufficient for study by the methylation method. In the methanolzate of the fully methylated disaccharide A, gas-liquid chromatography showed the presence of the methyl glycosides of 2,3,4-tri-O-methyl-D-xylose and 2,4-di-O-methyl-L-fucose. Thus, disaccharide A is 3-O-D-xylopyranosyl-L-fucose.

TABLE 1

Oligo-sac-charide	R _{Gal}	Monosaccharide composition*	Reducing terminal residue	Reaction with	
				TTC†	diphenylamine
A	1,23	Xyl, Fuc	Fuc	+	—
B	1,13	Fuc, Xyl	Xyl	—	—
C	0,90	Fuc, Gal	Gal	—	—
D	0,77	Xyl, Gal	Gal	+	+
E	0,47	Xyl, Fuc, Gal	Gal	—	—
F	0,19	Gal	Gal	—	—

* Xyl = D-xylose; Fuc = L-fucose; Gal = D-galactose.

† TTC = triphenyltetrazolium chloride.

Institute of Biologically Active Substances, Far-Eastern Scientific Center. Translated from *Khimiya Prirodnykh Soedinenii*, No. 1, pp. 107-108, January-February, 1973. Original article submitted July 12, 1972.

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The methanolyzate of disaccharide D was found to contain the methyl glycosides of 2,3,4-tri-O-methyl-D-xylose and 2,3,6-tri-O-methyl-D-galactose. Consequently, disaccharide D is 4-O-D-xylopyranosyl-D-galactose.

Thus, the side chains of sargassans contain the following oligosaccharide fragments: Xyl 1→3 Fuc; Fuc 1→2 Xyl; Fuc 1→2 Gal; Xyl 1→4 Gal; Xyl-Fuc 1→2 Gal; (Gal)_n1→2 Gal.

LITERATURE CITED

1. Yu. S. Ovodov, V. A. Khomenko, and T. F. Guseva, *Khim. Prirodn. Soedin.*, 285 (1970).
2. V. A. Khomenko, A. F. Pavlenko, T. F. Solov'eva, and Yu. S. Ovodov, *Khim. Prirodn. Soedin.*, 393 (1971).
3. V. M. Easterwood and B. J. L. Huff, *Svensk Papperstidn.*, 72, 768 (1969).
4. G. Avigad, R. Zelinson, and S. Hestrin, *Biochem. J.*, 80, 57 (1961).
5. S. Schwimmer and A. Bevenue, *Science*, 123, 543 (1956).
6. T. J. Painter, *Chem. Ind. (London)*, 34 (1963).